## Mass Spectral Fragmentation Pattern of 3-Methyl- and 3-Phenyl-4-(N-methylarylhydrazono)isoxazol-5-ones

Neil G. Keats, Shyam K. Singh and Lindsay A. Summers\*

Department of Chemistry, The University of Newcastle, 2308, New South Wales, Australia Received December 11, 1985

3-Methyl-4-(N-methylarylhydrazono)isoxazol-5-ones and 3-phenyl-4-(N-methylarylhydrazono)isoxazol-5-ones undergo considerable fragmentation on electron impact involving rupture of the isoxazolone ring and bonds in the N-methylarylhydrazono side chain.

## J. Heterocyclic Chem., 23, 1119 (1986).

It was recently reported [1] that methylation of 3-methyl-4-arylhydrazonoisoxazol-5-ones (I) with methyl iodide gave a mixture of 2,3-dimethyl-4-arylazoisoxazol-5-ones (II) and 3-methyl-4-(N-methylarylhydrazono)isoxazol-5-ones (IV) likewise afforded 2-methyl-3-phenyl-4-arylazoisoxazol-5-ones (V) and 3-phenyl-4-(N-methylarylhydrazono)isoxazol-5-ones (VI). We have already discussed the mass spectral fragmentation pattern [2] of 2,3-dimethyl- and 2-methyl-3-phenyl-4-arylazoisoxazol-5-ones and have summarized other work on the mass spectra of isoxazol-5-ones. This paper is concerned with the mass spectral fragmentation patterns of some 3-methyl-4-(N-methylarylhydrazono)isoxazol-5-ones

$$CH_{3} - C - C = N - NH - R$$

$$I$$

$$CH_{3} - C - C - N - N - R$$

$$CH_{3} - C - C - N - N - R$$

$$CH_{3} - C - C - N - N - R$$

$$III$$

$$CH_{3} - C - C - N - N - R$$

$$III$$

$$C_{6}H_{3} - C - C - N - N - R$$

$$CH_{3} - C - C - N - N - R$$

$$CH_{3} - C - C - N - N - R$$

$$CH_{3} - C - C - N - N - R$$

$$CH_{3} - C - C - N - N - R$$

$$CH_{3} - C - C - N - N - R$$

$$CH_{3} - C - C - C - N - N - R$$

$$CH_{3} - C - C - C - N - N - R$$

VΙ

and 3-phenyl-4-(N-methylarylhydrazono)isoxazol-5-ones.

The electron impact mass spectra of seven 3-methyl-4-(N-methylarylhydrazono)isoxazol-5-ones, namely 3-methyl-4-(N-methylphenylhydrazono)isoxazol-5-one (III, R = H), 3-methyl-4-(N-methyl-o-methylphenylhydrazono)isoxazol-5-one (III,  $R = 2-CH_3$ ), 3-methyl-4-(N-methyl-m-methylphenylhydrazono)isoxazol-5-one (III, R = 3-CH<sub>3</sub>), 3-methyl-4-(N-methyl-p-methylphenylhydrazono)isoxazol-5-one (III, R = 4-CH<sub>3</sub>), 3-methyl-4-(N-methyl-o-chlorophenylhydrazono)isoxazol-5-one (III, R = 2-Cl), 3-methyl-4-(Nmethyl-m-chlorophenylhydrazono)isoxazol-5-one (III, R = 3-Cl) and 3-methyl-4-(N-methyl-p-chlorophenylhydrazono)isoxazol-5-one (III, R = 4-Cl) and three 3-phenyl-4-(Nmethylarylhydrazono)isoxazol-5-ones, namely 3-phenyl-4-(N-methylphenylhydrazono)isoxazol-5-one (VI, R = H), 3-phenyl-4-(N-methyl-p-methylphenylhydrazono)isoxazol-5-one (VI, R = CH<sub>3</sub>) and 3-phenyl-4-(N-methyl-p-chlorophenylhydrazono)isoxazol-5-one (VI, R = Cl) were examined. Details of the principal peaks in the spectra are given in Tables 1 and 2 and the spectra of the parent compounds of the two series, 3-methyl-4-(N-methylphenylhydrazono)isoxazol-5-one (III, R = H) and 3-phenyl-4-(Nmethylphenylhydrazono)isoxazol-5-one (VI, R = H) are given in Figures 1 and 2 respectively by way of illustration. The synthesis of the ten compounds has been reported [1].

The mass spectra of the seven 3-methyl-4-(N-methylarylhydrazono)isoxazol-5-ones are interesting (Table 1). With the parent compound 3-methyl-4-(N-methylphenylhydrazono)isoxazol-5-one (III, R=H) the base peak is due to the  $C_6H_5^+$  ion at mass 77. The molecular ion  $C_{11}H_{11}N_3O_2^{++}$  at mass 217 gives a peak amounting to 29% of the intensity of the base peak. Fragmentation of the molecular ion proceeds along a number of routes involving either rupture of the isoxazolone ring or rupture of bonds in the N-methylphenylhydrazono side chain. Rupture of the isox

azolone ring with loss of  $CO_2$  and N affords a  $C_{10}H_{11}N_2^+$  ion of mass 159 (13%) as depicted in Scheme 1, while loss of  $CH_3CN$  and  $CO_2$  affords a  $C_8H_8N_2^{++}$  species of mass 132 (11%) depicted as a benzopyrazole derivative (Scheme 2). This latter ion can lose  $H^+$  to afford a  $C_8H_7N_2^+$  ion of mass 131 (8%) and lose  $CH_3^+$  to afford a  $C_7H_8N_2^+$  ion at mass 117 (9%) (Scheme 2). Rupture of the N-N bond of the N-methylphenylhydrazono side chain affords a  $C_7H_8N^+$  ion of mass 106 (56%) depicted as a benzaziridine derivative (Scheme 3). This ion may lose  $H^+$  to afford a  $C_7H_7N^{++}$ 

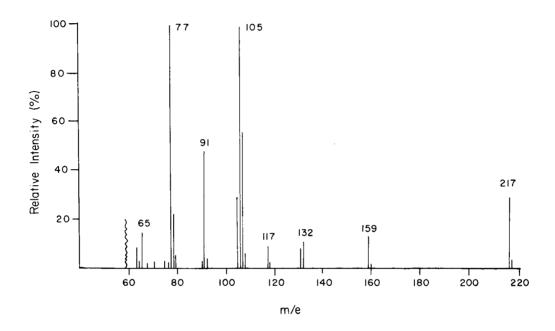


Figure 1. Mass spectrum of 3-methyl-4-(N-methylphenylhydrazono)isoxazol-5-one (III, R = H).

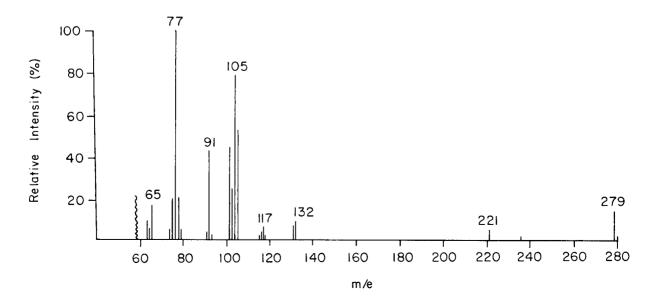


Figure 2. Mass spectrum of 3-phenyl-4-(N-methylphenylhydrazono)isoxazol-5-one (VI, R = H).

species which gives an intense peak at mass 105 (99%). Further loss of H accounts for the peak at mass 104 (29%), and loss of CH<sub>3</sub> gives the peak at mass 91 (48%) due to a C<sub>6</sub>H<sub>5</sub>N<sup>+</sup> species (Scheme 3). The only other peak

$$C_{e}H_{5}$$
  $C$   $C$   $N$   $N$   $N$   $CO$   $CH_{3}$   $CH_{3}$   $CI_{6}H_{13}N_{3}O_{2}$   $CI_{6}H_{13}N_{3}O_{2}$   $CI_{6}H_{5}$   $C$   $C$   $C$   $N$   $N$   $CI_{6}H_{13}N_{2}$   $CI_{15}H_{13}N_{2}$   $CI_{15}H_{13}N_{2}$ 

in the spectrum which deserves comment is the one due to the  $C_6H_6^{+}$  species at mass 78 (22%). This must arise by a hydrogen migration to the  $C_6H_5^{+}$  ion at mass 77 which is the base peak in the spectrum.

The spectra of the three 3-methyl-4-(N-methylmethylphenylhydrazono)isoxazol-5-ones (III,  $R=2\text{-}CH_3$ ), (III,  $R=3\text{-}CH_3$ ) and (III,  $R=4\text{-}CH_3$ ) can be explained in a similar way by fragmentation routes analogous to those in Schemes 1-3. The base peaks in the case of compound (III,  $R=2\text{-}CH_3$ ) and compound (III,  $R=3\text{-}CH_3$ ) are at mass 91 due to the respective tolyl ( $C_7H_7^+$ ) ions. With compound (III,  $R=4\text{-}CH_3$ ) the tolyl ion amounts to 96% of the base

peak which is due to the  $C_8H_{10}N^+$  ion at mass 120 (cf. Scheme 3). Peaks due to  $C_8H_{10}N^+$  ions are also prominent in the spectra of compound (III,  $R=2\text{-}CH_3$ ) (77%) and compound (III,  $R=3\text{-}CH_3$ ) (85%). As in the spectrum of the parent member of the series, compound (III, R=H), hydrogen migrations are invoked to explain some of the

Scheme

peaks, notably those due to  $C_8H_{11}N^{+}$  species at mass 121 and  $C_7H_8N^+$  ions at mass 106.

The spectra of the three 3-methyl-4-(N-methylchloro-

phenylhydrazono)isoxazol-5-ones (III, R=2-Cl), (III, R=3-Cl) and (III, R=4-Cl) are also readily explained along similar lines. The base peak in all three cases is due to a  $C_7H_6ClN^*$  species at mass 139 (cf. Scheme 3). It is interesting to note that rupture of the isoxazolone ring and the N-methylarylhydrazono side chain frequently occurs before loss of chlorine.

The spectra of the three 3-phenyl-4-(N-methylarylhydrazono)isoxazol-5-ones (VI, R = H), (VI,  $R = CH_3$ ) and (VI, R = Cl) are explained by fragmentation patterns similar to those in the 3-methyl-4-(N-methylarylhydrazono)isoxazol-5-one series (Table 2) although there are some interesting differences. With the parent member of the series compound (VI, R = H) the molecular ion  $C_{16}H_{13}N_3O_2^{++}$  gives a peak at mass 279 of 14% of the in-

Table 1

Empirical Formula of Fragment Ions [a] in the Mass Spectra of 3-Methyl-4-(N-methylarylhydrazono)isoxazol-5-ones III

m/e	Formula	R = H	$R = 2\text{-}CH_3$	$R = 3-CH_3$	Intensity (%) R = 4-CH <sub>3</sub>	R = 2-Cl	R = 3-Cl	R = 4-Cl
253	$C_{11}H_{10}^{37}ClN_3O_2$	-		•		-	6	12
251	$C_{11}H_{10}^{35}ClN_3O_2$	-	-	-	-	11	19	36
231	$C_{12}H_{13}N_3O_2$		26	28	45	-	-	•
217	$C_{11}H_{11}N_3O_2$	29		-		-	-	-
195	$C_{10}H_{10}^{37}ClN_2$	•	-			6	-	
193	$C_{10}H_{10}^{35}ClN_2$				-	21	12	13
173	$C_{11}H_{13}N_2$	-	8	5	9	-		-
166	C <sub>8</sub> H <sub>7</sub> <sup>35</sup> ClN <sub>2</sub>			•			-	16
165	$C_8H_6^{35}ClN_2$	-	•	-		-		6
159	$C_{10}H_{11}N_2$	11		•		-		
153	$C_7H_4^{37}ClN_2$			-	-		-	12
152	$C_7H_5^{35}ClN_2$		-	_	_		-	7
151	$C_7H_4^{35}CIN_2$	_			_		_	43
146	$C_9H_{10}N_2$		5	14	12		-	-
145	$C_9H_9N_2$	_	6	7	20	-	-	•
142	$C_7H_7^{37}CIN$	_			-	17	12	27
141	$C_7H_6^{37}CIN$	_		_	-	37	37	46
140	$C_7H_7^{35}CIN$	-	-	_		60	43	85
139	$C_7H_6^{35}CIN$	<u>-</u>	-		-	100	100	100
138	C <sub>7</sub> H <sub>5</sub> <sup>35</sup> ClN			-	-	21	17	26
132	$C_8H_8N_2$	11	-	-			-	
131	$C_8H_7N_2$	8	9	5	13	23	23	6
127				-	•	-	7	10
127	C <sub>6</sub> H <sub>4</sub> <sup>37</sup> ClN		-				26	27
	$C_6H_4$ <sup>35</sup> CIN $C_8H_{11}N$	-	13	41	- 29	•	-	-
121		-	13 77	85	100	•	-	-
120	C <sub>8</sub> H <sub>10</sub> N		67	67	82	-	-	-
119	C <sub>8</sub> H <sub>9</sub> N	•	96	27	23	-	•	-
118	C <sub>8</sub> H <sub>8</sub> N	-	8	-	- 23	•	-	•
117	C <sub>8</sub> H <sub>7</sub> N	-		-	•	-	-	•
114	C <sub>7</sub> H <sub>5</sub> N <sub>2</sub>	9	•	-	•	•		•
114	C <sub>6</sub> H <sub>5</sub> <sup>37</sup> Cl	•	-	•	•		5	-
113	C <sub>6</sub> H <sub>4</sub> <sup>37</sup> Cl	-	-	-	-	8	16	20
112	C <sub>6</sub> H <sub>5</sub> <sup>35</sup> Cl	-	•	•	-		17	-
111	C <sub>6</sub> H <sub>4</sub> <sup>35</sup> Cl		•		-	21	47	53
106	C <sub>7</sub> H <sub>8</sub> N	56	11	12	8	-		
105	C,H,N	99	16	41	30	8	15	21
104	C <sub>7</sub> H <sub>6</sub> N	29	15	14	14	6	15	12
101	C <sub>5</sub> H <sub>4</sub> <sup>37</sup> Cl	•	-	-	•	5		6
99	C <sub>5</sub> H <sub>4</sub> 35Cl	-		-	-	17	11	18
92	C <sub>7</sub> H <sub>8</sub>	-	15	19	19	-	•	•
91	C <sub>7</sub> H <sub>7</sub>	•	100	100	96	-	•	-
	C <sub>6</sub> H <sub>5</sub> N	48	•	-	-			-
90	C <sub>6</sub> H <sub>4</sub> N	•			•	11	23	13
89	C <sub>7</sub> H <sub>5</sub>	-	6	7	-	•	-	=
78	C <sub>6</sub> H <sub>6</sub>	22	24	19	15			•
77	C <sub>6</sub> H <sub>5</sub>	100	30	27	26	96	40	43
76	C <sub>6</sub> H <sub>4</sub>	-	•	-	-		7	7
75	C <sub>6</sub> H <sub>3</sub>	-	-	-	-	22	32	36
73	C <sub>3</sub> H <sub>2</sub> <sup>35</sup> Cl	-	•	-	•	9	7	10
67	$C_3H_3N_2$		-		•	9	8	7
65	C <sub>s</sub> H <sub>s</sub>	15	47	36	32	· 	•	-
63	C <sub>5</sub> H <sub>3</sub>	9	11	11	8	15	22	17

[a] Only those ions of mass >60 and of intensity ≥5% of the base peak are recorded. Peaks due to <sup>13</sup>C species are omitted from the table.

tensity of the base peak which is due to the  $C_6H_5^+$  ion at mass 77. Fragmentation of the molecular ion proceeds as in the 3-methyl series either by rupture of the isoxazolone ring or by rupture of bonds in the N-methylphenylhydra-

zono side chain. Rupture of the isoxazolone ring with loss of CO<sub>2</sub> and N affords a C<sub>15</sub>H<sub>13</sub>N<sub>2</sub><sup>+</sup> ion of mass 221 (5%) as shown in Scheme 4, while loss of C<sub>6</sub>H<sub>5</sub>CN and CO<sub>2</sub> affords a C<sub>8</sub>H<sub>8</sub>N<sub>2</sub><sup>+</sup> species of mass 132 (10%) depicted as before

Table 2

Empirical Formula of Fragment Ions [a] in the Mass Spectra of 3-Phenyl-4-(N-methylarylhydrazono)isoxazol-5-ones VI

		,	T . '. (0/)				
			Intensity (%				
m/e	Formula	R = H	$R = CH_3$	R = Cl			
215	C16H1237ClN3O2			7			
315	$C_{16}H_{12}^{-12}CIN_3O_2$ $C_{16}H_{12}^{-35}CIN_3O_2$	•	-	22			
313	$C_{17}H_{15}N_3O_2$	-	46				
293		14	40	-			
279	$C_{16}H_{13}N_3O_2$	14		•			
235	C <sub>16</sub> H <sub>15</sub> N <sub>2</sub>	-	6	-			
221	C <sub>15</sub> H <sub>13</sub> N <sub>2</sub>	5	-				
166	C <sub>6</sub> H <sub>7</sub> 35ClN <sub>2</sub>	•	-	16			
153	C <sub>7</sub> H <sub>4</sub> <sup>37</sup> ClN <sub>2</sub>	•	•	8			
151	C <sub>7</sub> H <sub>4</sub> 35ClN <sub>2</sub>	•	-	22			
146	C <sub>9</sub> H <sub>10</sub> N <sub>2</sub>	-	15	-			
145	C <sub>9</sub> H <sub>9</sub> N <sub>2</sub>	•	21	•			
143	C <sub>9</sub> H <sub>7</sub> N <sub>2</sub>	•	5	•			
	C <sub>7</sub> H <sub>8</sub> <sup>37</sup> ClN	•	-	10			
142	C <sub>7</sub> H <sub>7</sub> <sup>37</sup> ClN	-	-	25			
141	C <sub>7</sub> H <sub>8</sub> 35ClN	-	-	45			
140	C <sub>7</sub> H <sub>7</sub> 35CIN	-	-	72			
139	C7H635CIN	-	-	54			
138	C7H535CIN	-	•	15			
132	$C_8H_8N_2$	10	-	•			
131	$C_8H_7N_2$	7	14	•			
129	C <sub>8</sub> H <sub>5</sub> N <sub>2</sub>	•	6				
127	C <sub>6</sub> H <sub>4</sub> 37ClN	-	-	5			
125	C <sub>6</sub> H <sub>4</sub> 35ClN	-	-	19			
121	C <sub>8</sub> H <sub>11</sub> N	•	14	•			
120	C <sub>8</sub> H <sub>10</sub> N		100				
118	C <sub>8</sub> H <sub>8</sub> N	_	20				
117	C <sub>8</sub> H <sub>7</sub> N	-	5				
	C <sub>7</sub> H <sub>5</sub> N <sub>2</sub>	7					
116	C <sub>8</sub> H <sub>6</sub> N		5				
113	C <sub>6</sub> H <sub>4</sub> <sup>37</sup> Cl	_		11			
111	C <sub>6</sub> H <sub>4</sub> <sup>35</sup> Cl	-	_	32			
106	C <sub>7</sub> H <sub>8</sub> N	53	7				
105	C <sub>7</sub> H <sub>2</sub> N	78	44	12			
103	C <sub>7</sub> H <sub>6</sub> N	22	17	16			
	C <sub>7</sub> H <sub>5</sub> N	46	42	100			
103				6			
101	C <sub>5</sub> H <sub>4</sub> <sup>37</sup> Cl	•	•				
99	C₅H₄³⁵Cl	-		15			
92	C,H,	-	22	•			
91	C <sub>6</sub> H <sub>5</sub> N	44		•			
	C,H,	•_	88	•			
90	C <sub>6</sub> H <sub>4</sub> N	5		6			
	C7H6	-	5	-			
89	C <sub>7</sub> H <sub>5</sub>	•	6	•			
78	C <sub>6</sub> H <sub>6</sub>	21	16	•			
77	C <sub>6</sub> H <sub>5</sub>	100	37	49			
76	C <sub>6</sub> H₄	20	13	33			
75	C <sub>6</sub> H <sub>3</sub>	6	7	29			
74	C <sub>6</sub> H <sub>2</sub>	•	•	7			
73	C <sub>3</sub> H <sub>2</sub> 35Cl	•	-	6			
65	$C_5H_5$	18	41	-			
64	C <sub>5</sub> H <sub>4</sub>	6	•	•			
63	C <sub>5</sub> H <sub>3</sub>	10	12	13			

<sup>[</sup>a] Only those ions of mass >60 and of intensity ≥5% of the base peak are recorded. Peaks due to <sup>13</sup>C species are omitted from the table.

as a benzopyrazole derivative (Scheme 5). This species can lose H to afford a C<sub>8</sub>H<sub>7</sub>N<sub>2</sub> ion of mass 131 (7%) and lose CH<sub>3</sub> to afford a C<sub>7</sub>H<sub>5</sub>N<sub>2</sub> ion at mass 117 (7%). The benzonitrile molecular ion C<sub>7</sub>H<sub>5</sub>N<sup>+</sup> at mass 103 (46%) gives rise to a prominent peak in the spectrum (Scheme 6). The corresponding fragmentation to afford the acetonitrile molecular ion C<sub>2</sub>H<sub>3</sub>N<sup>+</sup> at mass 41 was not noticed in the spectrum of 3-methyl-4-(N-methylphenylhydrazono)isoxazol-5-one (III, R = H) possibly due to the low mass of the acetonitrile molecular ion. Rupture of the N-N bond of the N-methylphenylhydrazono side chain of 3-phenyl-4-(N-methylphenylhydrazono)isoxazol-5-one (VI, R = H) affords a C<sub>7</sub>H<sub>8</sub>N<sup>+</sup> ion at mass 106 (53%) depicted as a benzaziridine derivative (Scheme 7). This ion may lose H to afford a C<sub>7</sub>H<sub>7</sub>N<sup>+</sup> species at mass 105 (78%). Further loss of H accounts for the peak at mass 104 (22%) and loss of CH<sub>3</sub> gives the peak at mass 91 (44%) due to a C<sub>6</sub>H<sub>5</sub>N<sup>+</sup> species (Scheme 7). The C<sub>6</sub>H<sub>5</sub>N<sup>+</sup> species may lose H to give a C<sub>6</sub>H<sub>4</sub>N<sup>+</sup> ion at mass 90 (5%). The only other peak in the spectrum which deserves comment is the one due to the C<sub>6</sub>H<sub>6</sub><sup>+</sup> species at mass 78 (21%). This species must arise, as in the 3-methyl series, by a hydrogen migration.

The spectrum of 3-phenyl-4-(N-methyl-p-methylphenylhydrazono)isoxazol-5-one (VI, R = CH<sub>3</sub>) can be explained by fragmentation routes analogous to those shown in Schemes 4-7. The base peak is at mass 120 due to a  $C_8H_{10}N^+$  ion analogous to the  $C_7H_8N^+$  ion shown in Scheme 7. The benzonitrile molecular ion  $C_7H_5N^+$  at mass 103 in this case results in a peak of 42% of the intensity of the base peak. The p-tolyl ion,  $C_7H_7^+$  at mass 91 gives rise to a prominent peak in the spectrum (88%).

The spectrum of 3-phenyl-4-(N-methyl-p-chlorophenyl-hydrazono)isoxazol-5-one (VI, R = Cl) can be explained in a similar way. In this case the base peak is at mass 103 due to the benzonitrile molecular ion  $C_7H_5N^{+-}$  (cf. Scheme 6). As before it is interesting to note that rupture of the isoxazolone ring and the N-methylarylhydrazono side chain frequently occurs before loss of chlorine and hydrogen migrations have sometimes to be invoked to explain a few of the peaks.

## **EXPERIMENTAL**

The spectra were determined with an A. E. I. MS-30 mass spectrometer. The samples were analysed by a direct insertion probe at an ionizing current of 70 eV. The ion source temperature was 200°.

The 3-methyl- and 3-phenyl-4-(N-methylarylhydrazono)isoxazol-5-ones were analytically pure [1].

## REFERENCES AND NOTES

<sup>[1]</sup> S. K. Singh and L. A. Summers, J. Heterocyclic Chem., 22, 457 (1985).

<sup>[2]</sup> N. G. Keats, S. K. Singh and L. A. Summers, J. Heterocyclic Chem., 22, 1531 (1985).